This Question Paper contains 12 Printed Pages.

Science પ્રશ્ન પેપરનો Set No. Question Paper

Sl. No. 904008

N-11 (E)

(MARCH, 2012)

PART - A

Time: 75 minutes]

[Maximum Marks: 50

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Instructions:

- **(1)** There are **50** objective type questions in this part and **all** are **compulsory**.
- (2)The questions are serially numbered from 1 to 50 and each carries 1 mark.
- (3)You are supplied with separate OMR sheet with the alternatives (A) \bigcirc , (B) \bigcirc , (C) O, (D) O against each question number. For each question, select the correct alternative and darken the circle ○ as • completely with the pen against the alphabet corresponding to that alternative in the given OMR sheet.
- From the following 1 to 50 questions, select the correct alternative from the given four answers and darken the circle with pen against the alphabet, against the number in OMR sheet.
- Each question carries 1 mark.
- 1. Who invented the simple battery first?
 - (A) Faraday

Ohm **(B)**

(C) Volta

- (D) Alva Edison
- 2. Which of the following oxide is not a neutral oxide?
 - (A) CO

(B) N_2O

(C) H_2O

- (**D**) SO₂
- 3. What is IUPAC name of Acetone?
 - (A) Propanal

(B) Propanone

(C) Propanol

- (**D**) Propanoic acid
- 4. What is the diameter of nano-shells which are attached only to Cancerous cells?
 - (A) 400 nm

(B) 200 nm

(C) 100 nm

50 nm (\mathbf{D})

			[2]			
5.	Which metal causes Minamata disease?					
	(A)	Copper	(B)	Lead		
	(C)	Manganese	(D)	Mercury		
6.	What is the chemical formula of Silver glance?					
	(A)	AgCl	(B)	Ag_2S		
	(C)	${ m SiO}_2$	(D)	${\rm AgNO}_3$		
7.	Whi	ch organic compounds contain	«∩]	H functional group?		
••		Carboxylic acid	(B)	Ketone		
		Aldehyde	(D)	•		
8.	Wh	at is the atomic number of Tra	, mailman	oia alamanta ?		
0.		Z = 92	nsurai (B)			
	, ,	Z = 92 $Z > 92$. ,	$Z \leq 92$ $Z \leq 92$		
9.	Give	e the name of theory proposed	by Eri	nst Haeckel.		
	(A)	Theory of germplasm	(B)	Theory of mutation		
	(C)	Theory of recapitulation	(D)	Theory of natural selection		
10.	Which space shuttle met with an accident at the time of landing, in which Indian Astronaut Kalpana Chawla died in February 2003?					
	(A)	Colombia	(B)	Challenger		
	(C)	Discovery	(D)	PSLV		
11.	. Which compound is condensation polymer?					

11. Which compound is condensation polymer?

(A) Nylon

(B) PVC

(C) Natural Rubber

(D) Teflon

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(A)

 CH_3OH

13. Which structure is developed in the wall of Uterus to provide nutrition to foetus?

[3]

(A) Amnion

(B) Fallopian tube

(C) Umbilical cord

(D) Placenta

14. Pons connects which two organs with the help of transverse band of nerves?

- (A) Nerves of brain and spinal cord.
- (B) Both the cerebral hemispheres.
- (C) Cerebellum and Central nervous system.
- (D) Sympathetic and Parasympathetic.

15. Which of the following hormones is responsible for shedding of leaves in plants?

(A) Abscisic acid

(B) Gibberellin

(C) Cytokinin

(D) Auxin

16. Which metal of the following metals is more active?

(A) Mg

(B) Zn

(C) Ca

(D) Al

17. In which plant, vegetative propagation by leaf takes place?

(A) Sweet potato

(B) Potato

(C) Phalsa

(D) Bryophyllum

18. What is the percentage of Carbon in hard steel?

(A) 0.1 to 0.4

(B) 1.5 to 2.5

(C) 2.5 to 3.5

(D) 0.5 to 1.5

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19.	When current passes	through a conductor,	in which	direction	magnetic
	field is produced?			•	

- (A) In a direction of current.
- (B) In the opposite direction of current.
- (C) Circular around the conductor.
- (D) Perpendicular to the direction of current.
- 20. Which organism normally shows asexual reproduction by fragmentation?
 - (A) Oscillatoria

(B) Amoeba

(C) Paramoecium

- (D) Penicillium
- 21. Which material provides the mechanical support to cells of trachea?
 - (A) Prothrombin

(B) Cellulose

(C) Lignin

- (D) Pectin
- 22. How many Light-year away, the Sun is located from the galactic centre?
 - (A) 250

(B) 30,000

(C) 22.5

- (D) 15,000
- 23. In which material, Sulphur is soluble?
 - (A) Carbon disulphide
- (B) Bromine

(C) Heavy water

- (D) Distilled water
- 24. Give the chemical formula of Calcium sulphate hemihydrate.
 - (A) CaSO₄·2H₂O

(B) $CaSO_4 \cdot \frac{1}{2}H_2O$

(C) CaSO₄·10H₂O

(D) CaSO₄·H₂O

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- 25. Which pair of the following is complementary colours?
 - (A) Blue and yellow

- (B) Green and yellow
- (C) Red and magenta
- (D) Blue and magenta
- 26. What is the formula for Electric Power?
 - (A) $P = I^2Rt$

 $(\mathbf{B}) \qquad \mathbf{P} = \frac{\mathbf{W}}{t}$

(C) $P = VI \times t$

- (D) P = VG
- 27. What is the name of finger like projections in the small intestine of human?
 - (A) Vermiform Appendix
- (B) Villi

(C) Gizzard

- (D) Food vacuole
- 28. Who developed the treatment technique for industrial and sewage waste water?
 - (A) NEERI

(B) NACO

(C) WHO

- (D) ISRO
- 29. Which metal oxide is used to obtain blue coloured glass?
 - (A) Ferric oxide

- (B) Chromium oxide
- (C) Manganese oxide
- (D) Cobalt oxide
- 30. Which mineral is necessary for blood clotting?
 - (A) Calcium

(B) Magnesium

(C) Phosphorus

- (D) Iron
- 31. By which, hormones are circulated?
 - (A) Water

(B) Nerve

(C) Blood

(D) Cytoplasm

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32.	Which catalyst is used for the industrial production of Hydrogen?							
	(A)	Iron	(B)	Nickel				
	(C)	Vanadium pentoxide	(D)	Palladium				
33.	Whi	ch satellite is launched by Indi	a for I	Direct to Home (DTH)?				
	(A)	INSAT - 4 A	(B)	IRS-1				
	(C)	METSAT	(D)	CARTOSAT				
34.	Whi	ch compound is used for bleach	ing cl	oths in laundry?				
	(A)	Bleaching Powder	(B)	Washing Powder				
	(C)	Baking Powder	(D)	Plaster of Paris				
35.	Whi	ch planet has atmosphere up to	1% of	the atmosphere of the Earth?				
	(A)	Mars	(B)	Venus				
	(C)	Jupiter	(D)	Saturn				
36.	Give	e the unit of rate of reaction.						
	(A)	Molar	(B)	Second				
	(C)	Second/Molar	(D)	Molar/Second				
37.	Which plant shows thigmonastic response?							
	(A)	Sunflower	(B)	Mimosa				
	(C)	Periwinkle	(D)	Bryophyllum				
22	****	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1						
38.		ch chromosome has satellite?	(20)	•••				
	(A)	Telocentric	(B)	Metacentric				
	(C)	Acrocentric	(D)	Sub-metacentric				

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39. Which scientist gave the law of Active mass	39.	Which	scientist	gave	the l	law	of	Active	mass	1
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- (A) Goldberg and Waag.
- (B) Lowry and Bronsted.
- (C) Boyle and Arrhenius.
- (D) Lewis and Sorensen.

40. Who gave the principle of Electromagnetic induction?

(A) Volta

10

(B) Ampere

(C) Faraday

(D) Oersted

- (A) Copper and Zinc
- (B) Copper and Tin
- (C) Nickel and Chromium
- (D) Lead and Tin

(A) Red

(B) Yellow

(C) Violet

(D) Blue

43. If 0.3 A current passes through a lamp, how many electrons will pass in 60 seconds?
$$(e = 1.6 \times 10^{-19})$$

(A) 2.88×10^{20}

(B) 1.125×10^{20}

(C) 2.25×10^{20}

(D) 1.8×10^{20}

- (A) Electric generator
- (B) Solenoid

(C) Electric motor

(D) Electric iron

(A) (B) (C) (D) How temp	ch organs perform the same fur Homologus organs. Analogus organs. Vestigial organs. Structurally homogeneous organs organs organs. The much the temperature of Separature of the body?	gans. crotum			
(B) (C) (D) How temp	Analogus organs. Vestigial organs. Structurally homogeneous organs. much the temperature of Separature of the body?	erotum	n in male is lower than the		
(C) (D) How tem; (A)	Vestigial organs. Structurally homogeneous orgonized much the temperature of Seperature of the body?	erotum	n in male is lower than the		
(D) How tem (A)	Structurally homogeneous orgonically much the temperature of Seperature of the body?	erotum	n in male is lower than the		
How tem	w much the temperature of Seperature of the body?	erotum	n in male is lower than the		
tem	perature of the body?		n in male is lower than the		
	0° C	(D)			
	-	(B)	3° C		
(C)	34° C	(D)	37° C		
Rays of light are entering from glass to glycerine. If refractive index of glass and glycerine are respectively 1.5 and 1.47, find the refracting index of glycerine with respect to glass.					
(A)	0.03	(B)	1.02		
(C)	2.20	(D)	0.98		
	_	is fo	rmed of a person having		
(A)	On retina	(B)	Behind retina		
(C)	Infront of retina	(D)	On lens of eye.		
In S	ponges, which structure is used	nich structure is used for excretion ?			
(A)	Contractile vacuole	(B)	Flame cells		
(C)	Nephridia	(D)	Osculum		
At v	which depth, we get necessary t	emper	ature for OTEC in oceans?		
(A)	0 m to 20 m	(B)	100 m to 300 m		
(C)		(D)	700 m to 900 m		
	(C) Raylof glinde (A) (C) At far-s (A) (C) In S (A) (C) At v (A)	Rays of light are entering from glass of glass and glycerine are respective index of glycerine with respect to graph (A) 0.03 (C) 2.20 At which place in eye, image far-sightedness (hypermetropia)? (A) On retina (C) Infront of retina In Sponges, which structure is used (A) Contractile vacuole (C) Nephridia At which depth, we get necessary to (A) 0 m to 20 m	Rays of light are entering from glass to go of glass and glycerine are respectively 1.5 index of glycerine with respect to glass. (A) 0.03 (B) (C) 2.20 (D) At which place in eye, image is for far-sightedness (hypermetropia)? (A) On retina (B) (C) Infront of retina (D) In Sponges, which structure is used for extended to the structure of the structure is used for extended to the structure of the structure is used for extended to the structure of the structu		

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(MARCH, 2012)

PART - B

Tim	e : 2.00 Hours]	[Maximum Marks: 50
Inst	ructions :-	
<i>(i)</i>	There are total four sections in this part.	
(ii)	All questions are compulsory.	
(iii)	Draw neat labelled diagram as per instructions.	
(iv)	There are internal options in some questions. Pay	attention to them.
(v)	Figures to the right indicate marks.	
	SECTION - A	
Give	short answer (in limit of 30 words) of the following	questions.
1.	At which temperature range; Petrol, Diesel, Keros	ene and lubricating
	oil is obtained in fractional distillation of Petroleu	m? 2
2.	What is Molarity? Give its unit.	2
	OR	,
2.	What is slow and fast reaction? Give example.	
3.	Why Carbon is important in development of Nano	-technology? 2
4.	Give the definition of Solar constant and its value	. 2
	OR	
4.	Write the name of various types of Coal and give t	he percentage of
	Carbon in each.	
5.	Give short information about Mercury.	. 2

[10]

SECTION - B

Write short answer (in the limit of 30 words) of the following questions.

6. Calculate the pH of 0.04 M aqueous solution of NaOH.

$$(\log_{10} 4 = 0.6021)$$

2

Write uses of Baking powder (NaHCO₃).

2

8. How pure water can be obtained by sewage treatment?

2

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9. Write the process of making Ethyl acetate by esterification with equation. 2

OR

- 9. Write difference between Soap and Detergent.
- 10. Give short information regarding Spinal cord of human.

2

SECTION - C

Write answers in short (in the limit of 50 words) of the following questions.

11. Explain the work of an Electric Generator with diagram.

3

12. Explain allotropes of Sulphur.

3

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- OR
- 12. Complete the following chemical reactions:

(i)
$$S_{(s)} + 2H_2SO_{4(aq)} \rightarrow$$

(ii)
$$SO_{2(g)} + 2H_2S_{(g)} \rightarrow$$

$$(iii) \ \ \mathrm{H_2S_2O_{7(l)} + H_2O_{(l)}} \rightarrow$$

13. Describe Erythrocyte (RBC) in short.

3

14. Explain Sex determination.

3

[11]

15. Explain Electroplating with suitable example.

3

OR

15. Explain series connections of Resistors and derive the formula of equivalent resistance.

SECTION - D

Write the answer of the following questions in detail (in the limit of 100 words).

16. Derive lens formula
$$\frac{1}{v} - \frac{1}{u} = \frac{1}{f}$$
.

5

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17. What is concentration or enrichment of Ore? Explain the method of enrichment of ore containing Sulphide with diagram.

5

OR

17. Explain Bayer's method for obtaining Alumina from Bauxite with equations.

18. Write short note: Aerobic respiration and Anaerobic respiration.

5

OR

18. What is Nutrition? Describe nutrition in Amoeba. (Draw diagram).